

Combustion of hydrogen (en-)rich(ed) fuel gases in gas turbines

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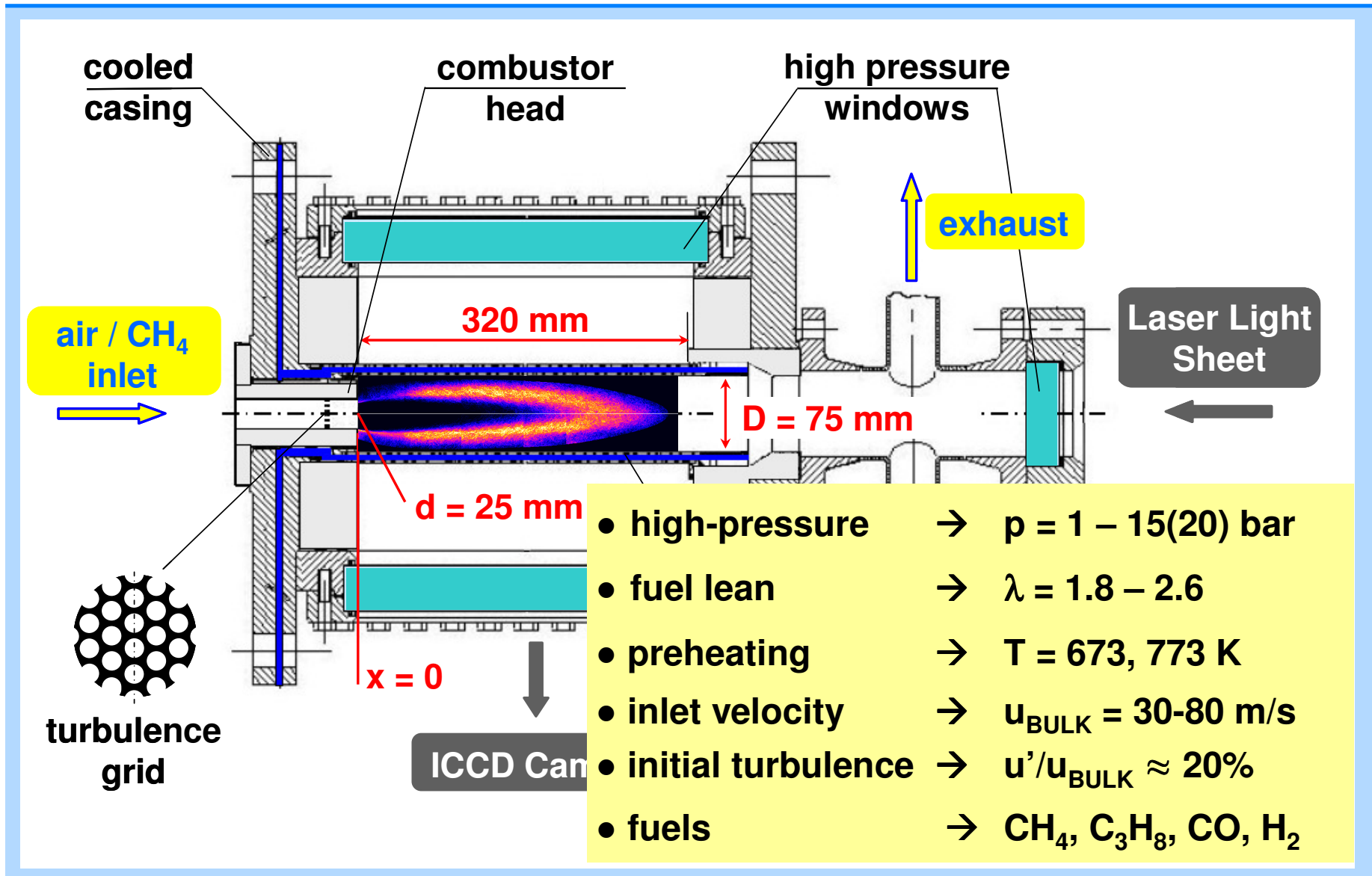
Motivation

- **improve lean premixed combustion technology for gas turbines**
 - improved flame stability (especially lean blow-out)
 - lower NO_x emissions
- **generation of experimental database for**
 - validation of numerical models
 - design rules for improved burner geometries

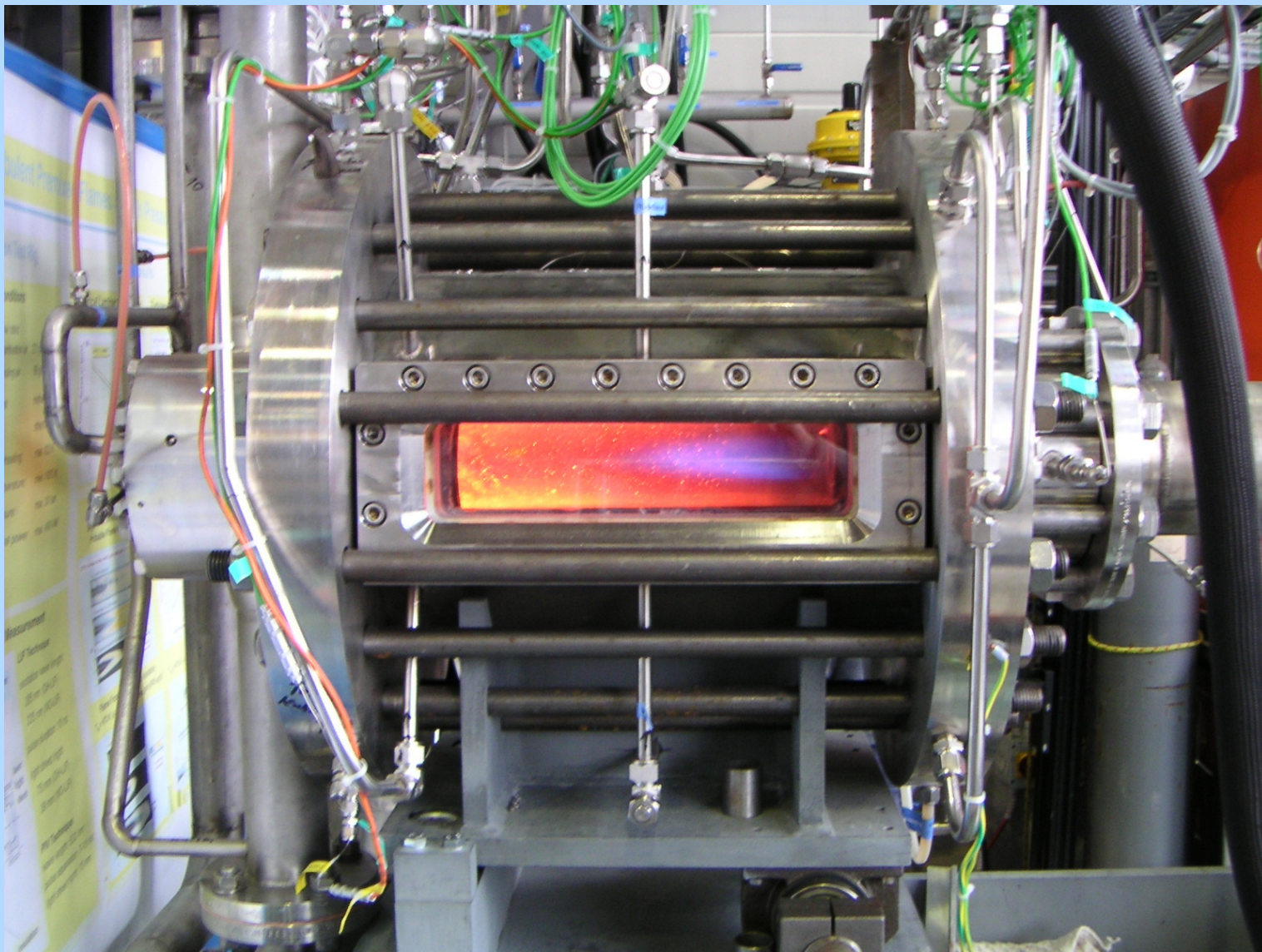
Objectives

- **flame characterisation for a broad fuel spectrum
(CH₄, CH₄/H₂, CH₄/C₃H₈, CO/H₂ (syngas), H₂ (>80%Vol.))
at gas turbine conditions (p: up to 20bar; $\Phi \approx 0.5$)**
 - influences of operating conditions and turbulence on flame structure (position, flame brush thickness, turbulent flame speed)
 - stability limits: Lean blowout (LBO), flashback
 - emissions: CO, NO_x

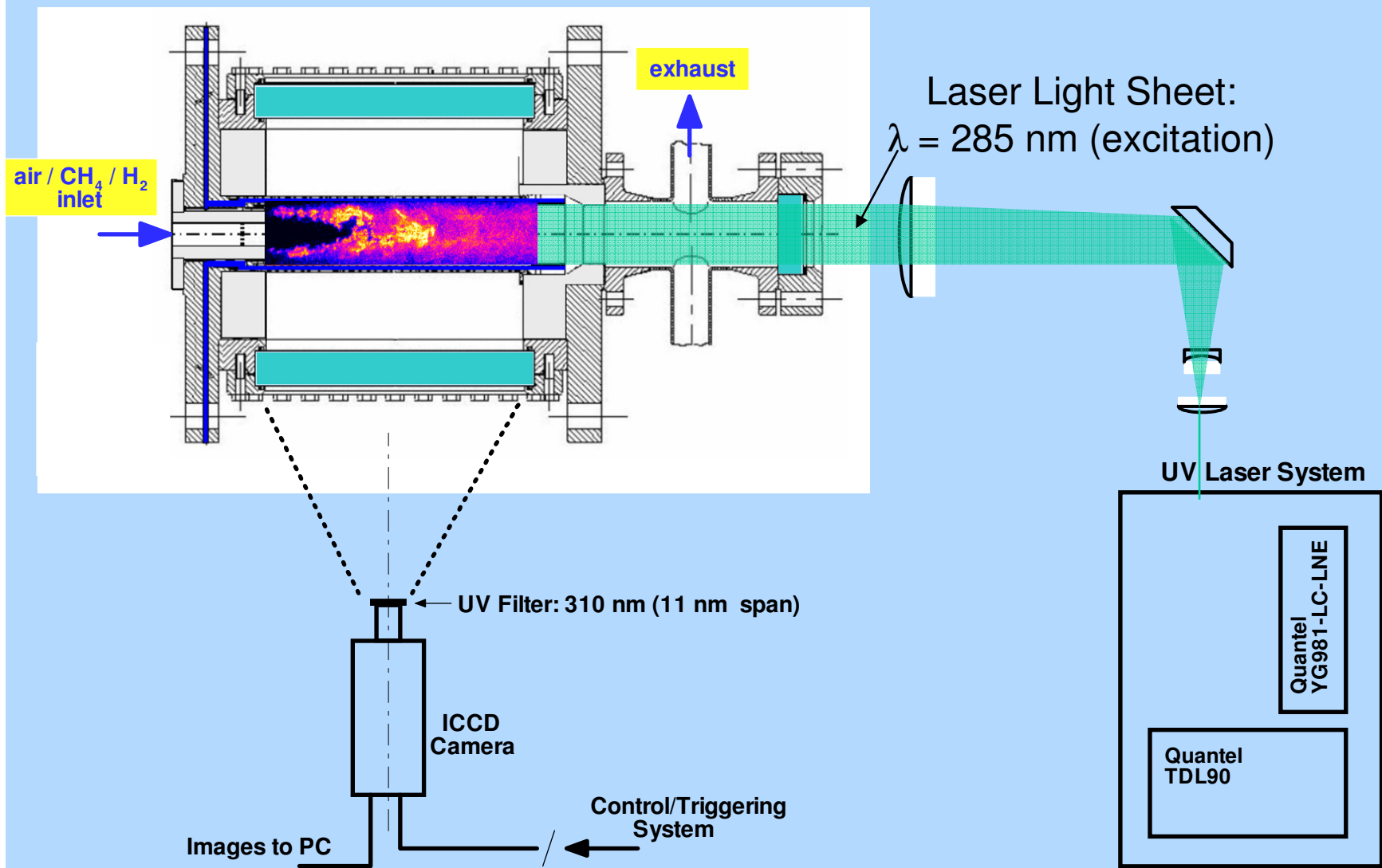
Experimental Set-up



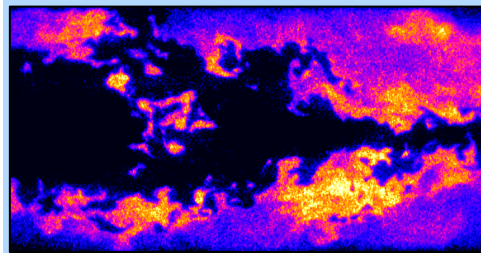
High pressure test rig



Experimental Setup: OH-PLIF



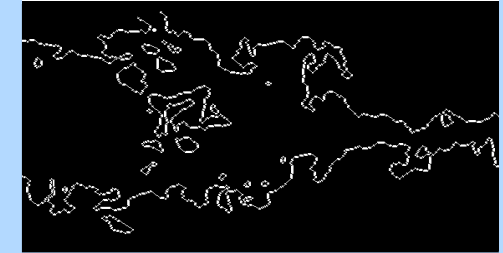
Flame Front Detection



Single shot OH-PLIF



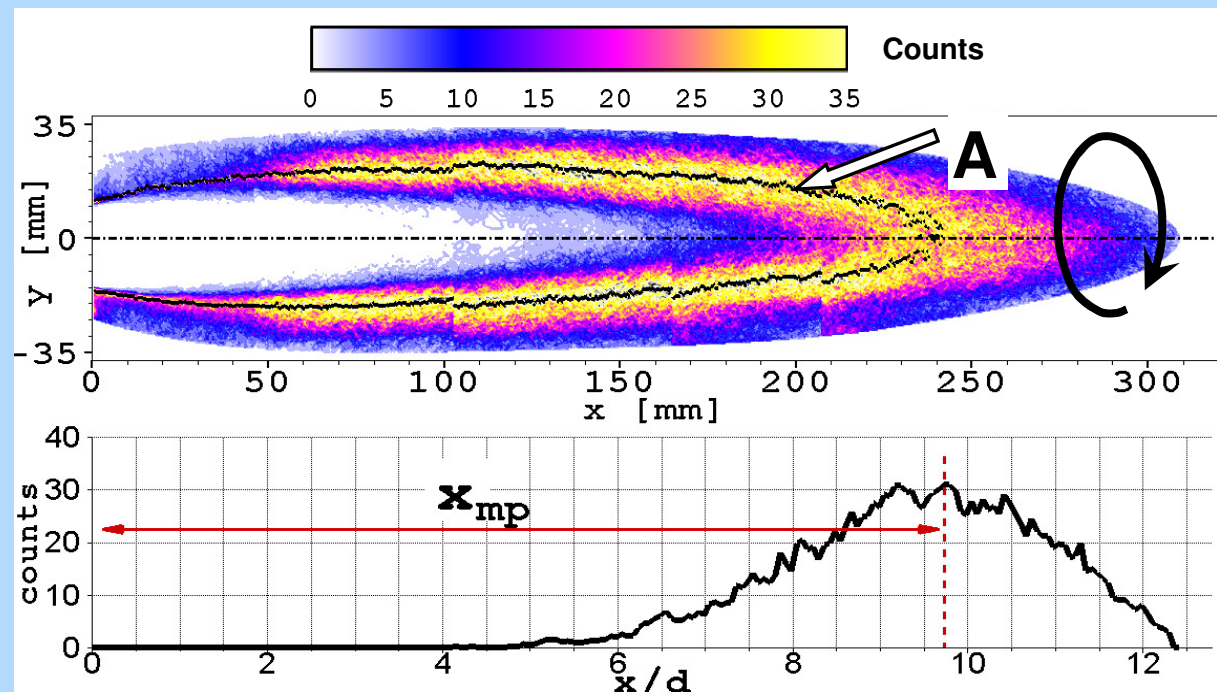
Binary image



Flame front contour

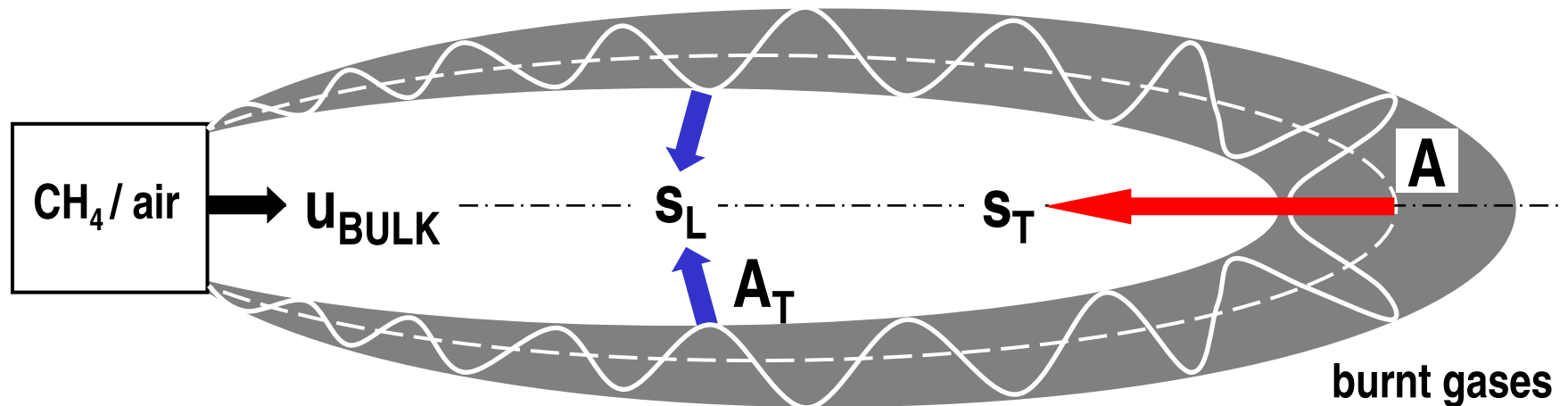
Turbulent flame
speed:

$$S_T = u_{\text{bulk}} \cdot A_{\text{inlet}} / A$$



Statistical analysis (800 single shots) → most probable flame front position x_{mp}

Turbulent Flame Speed



definition of s_T based on
mass continuity with
constant density
assumption

$$\rho_0 u_{\text{BULK}} A_0 = \rho_u s_T A = \rho s_L A_T$$

$$s_T = u_{\text{BULK}} \frac{A_0}{A}$$

$$\frac{s_T}{s_L} = \frac{A_T}{A}$$

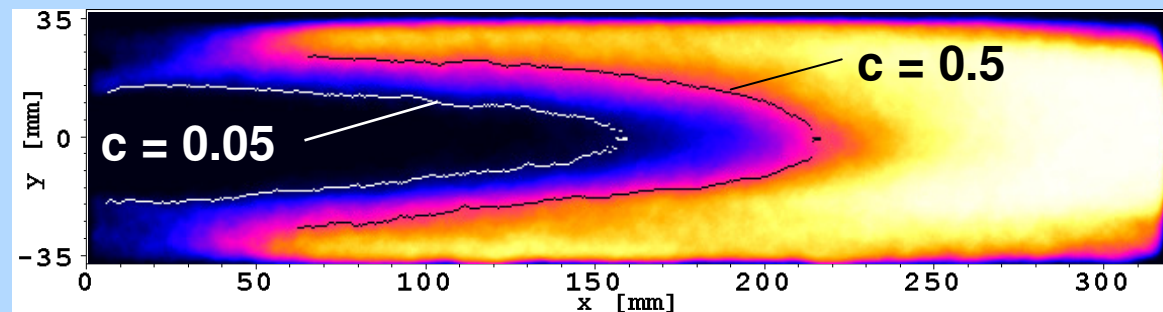
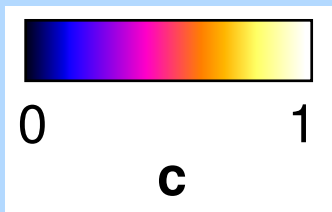
Turbulent Flame Speed S_T

Progress Variable Approach

progress variable definition:

(e. g. normalised temperature, mass fractions, here OH-PLIF Signal); u = unburnt, b = burnt

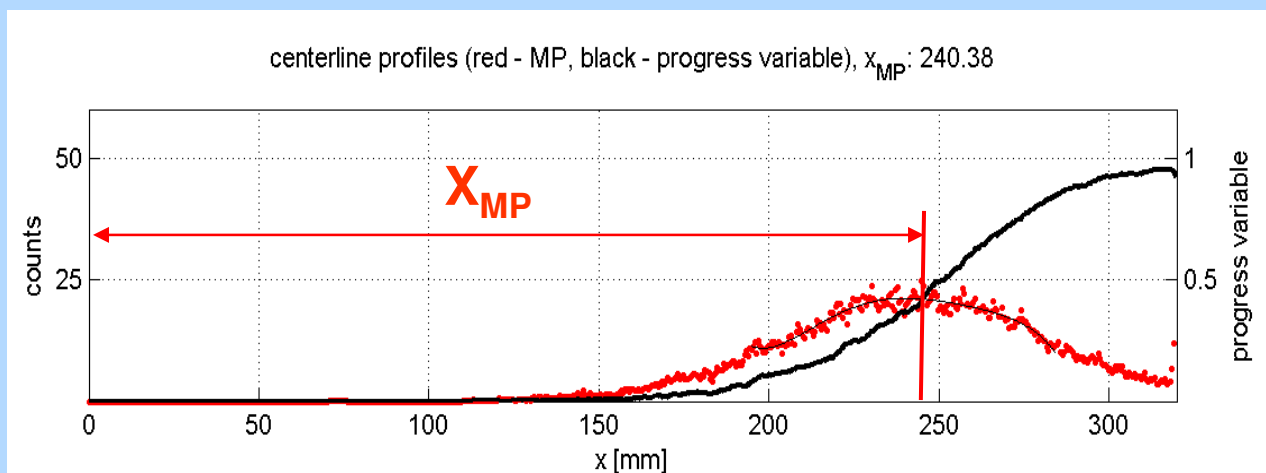
$$c = \frac{T(z) - T^u(z)}{T^b(z) - T^u(z)}$$



Turbulent flame speed:


$$S_T = u_{\text{bulk}} \cdot A_{\text{inlet}} / A$$

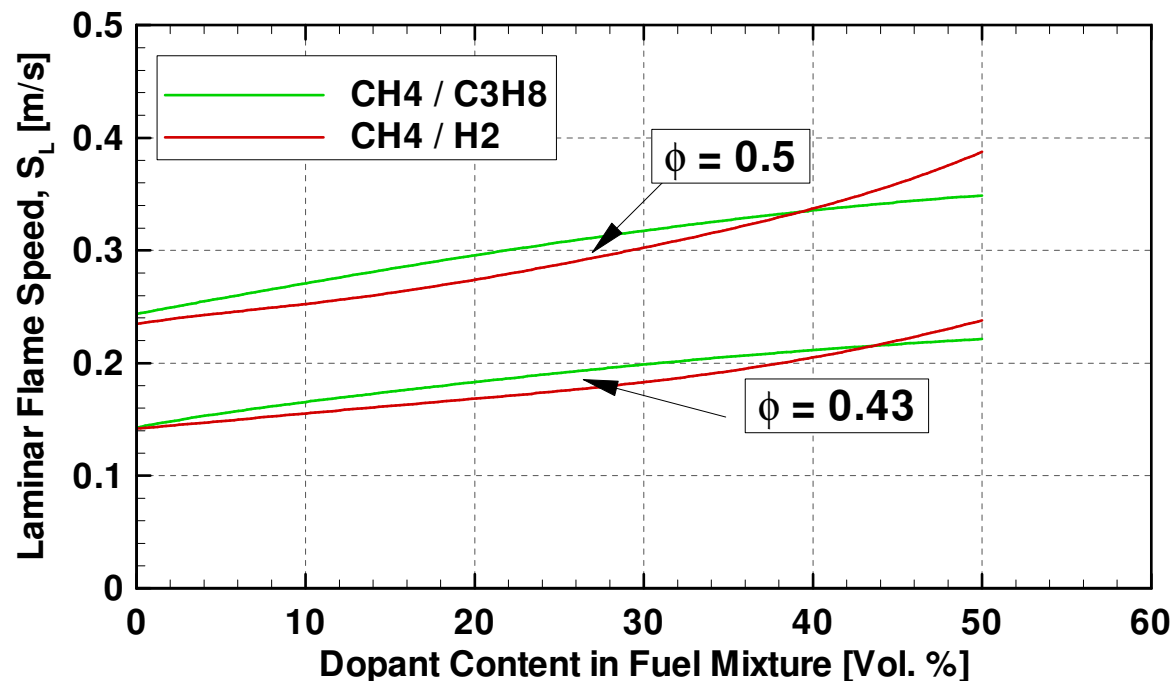
A calculated based on $c=0.05$



Focus

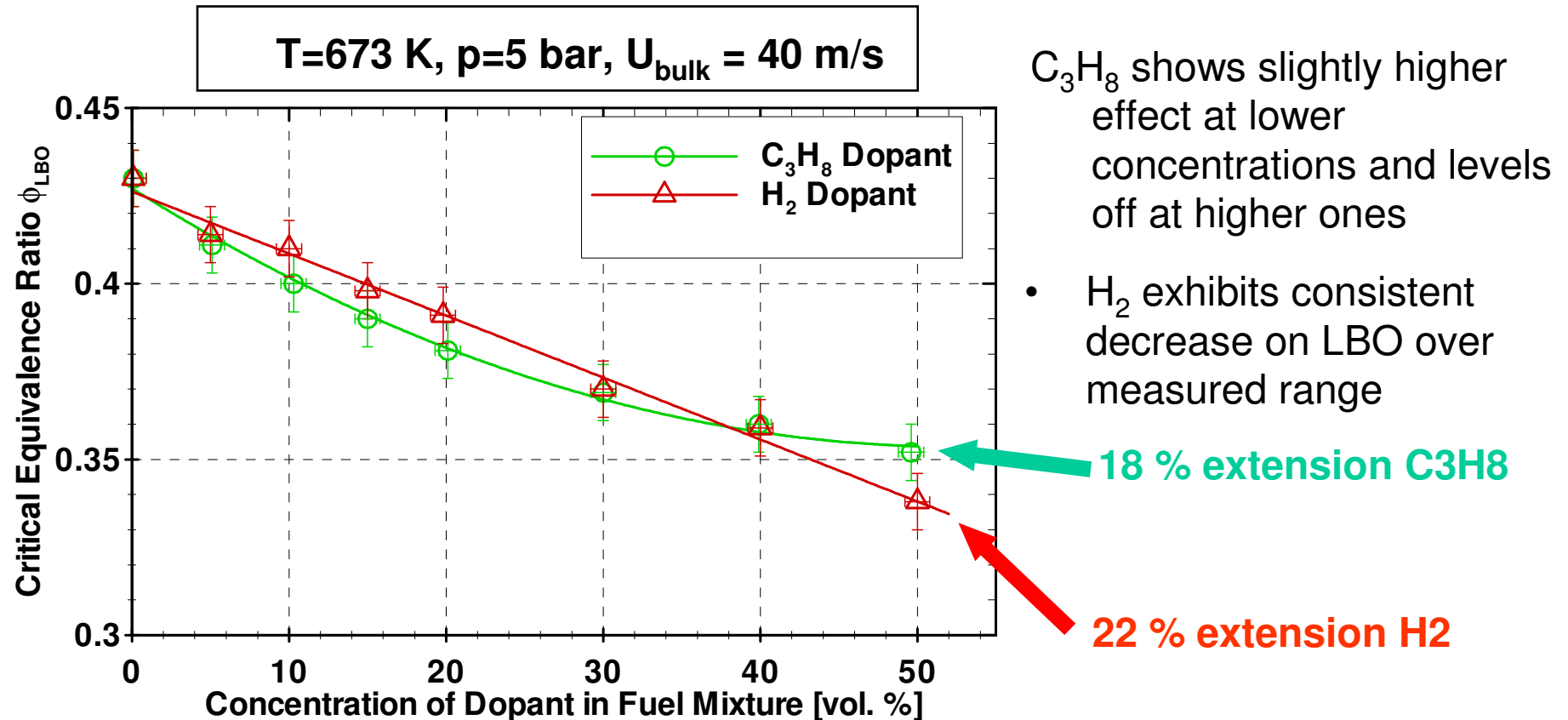
Enhanced Flame Stability (H₂-CH₄/air flames)

- Base Fuel is methane (CH_4)  main constituent of natural gas
- hydrogen (H_2) is added to enhance lean flame stability
 1. Highly reactive – high laminar flame speed
 2. Extremely light and diffusive
- higher hydrocarbons C_2+ represented by propane (C_3H_8)
 1. More reactive than methane - higher laminar flame speed
 2. Heavier molecule: lower diffusivity



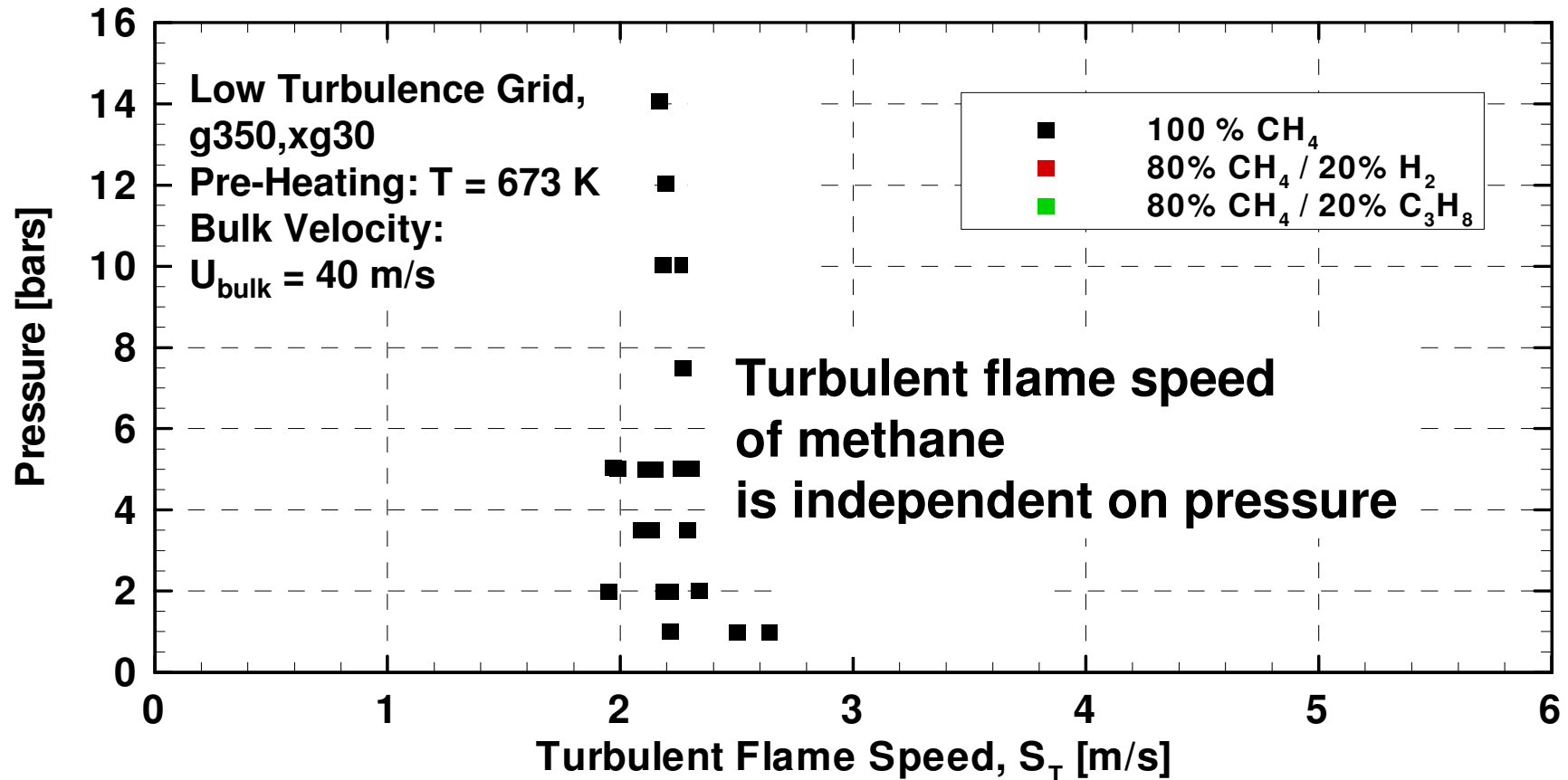
GRI-mech 3.0

Lean Blowout Limits (LBO): Influence of Fuel Blend



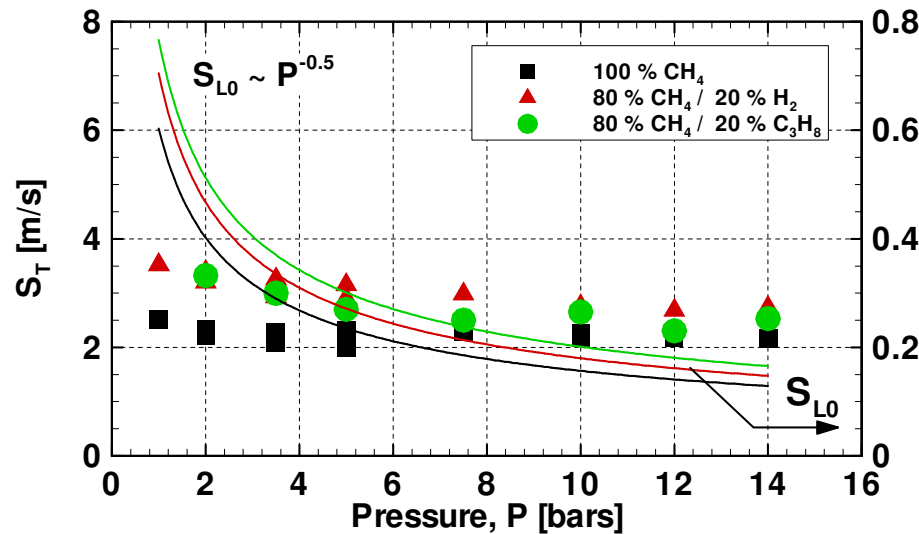
- Hydrogen results agree well with a study in a swirl stabilized combustor [1] which showed ~23 % extension at 50 % H₂ (Pressures 1, 4, 8 atm).

Turbulent Flame Speed



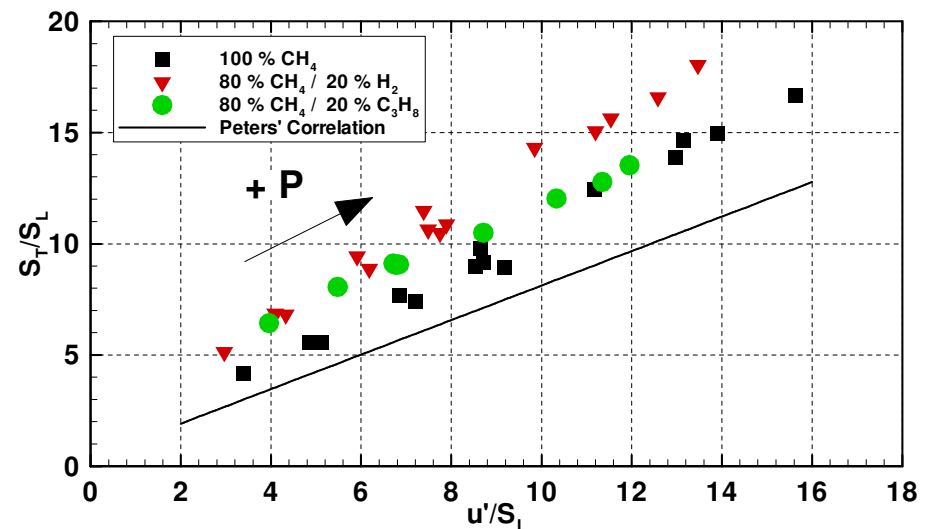
Turbulent flame speed of methane mixtures (with C_3H_8 , H_2) is dependent on pressure (even though the effect is very small above 10 bar)

Turbulent Flame Speed: Influence of Pressure



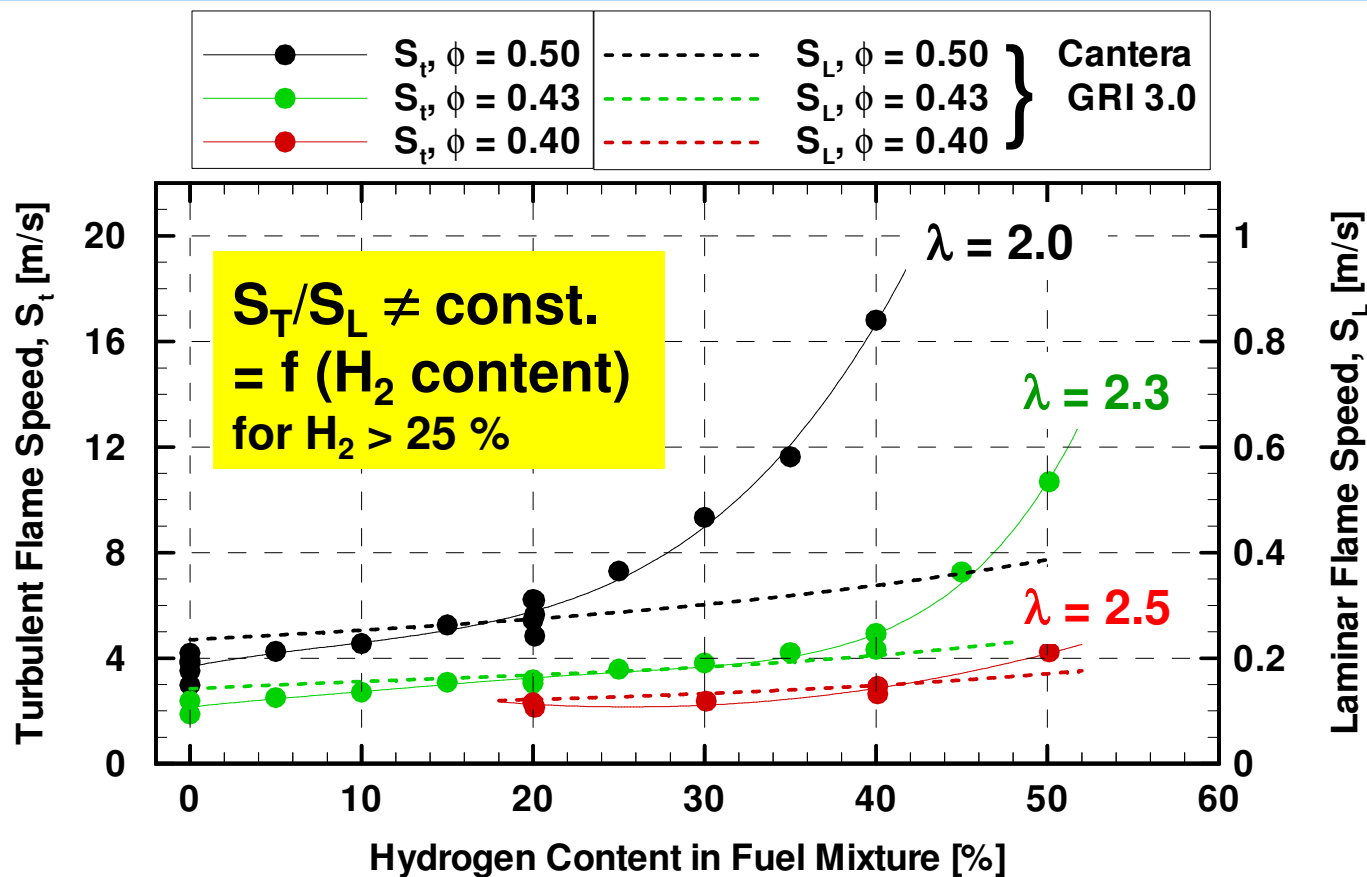
- Laminar flame speed is heavily dependent on pressure
- Turbulent flame speed is nearly independent of pressure for the three mixtures
- The increase in Re_T by increasing the pressure decreases the size of the smallest turbulent eddies

- The resulting increase in turbulent flame surface counteracts the decrease of laminar flame speed
- Confirms results of prior PhD work (P. Siewert) on pure methane/air flames



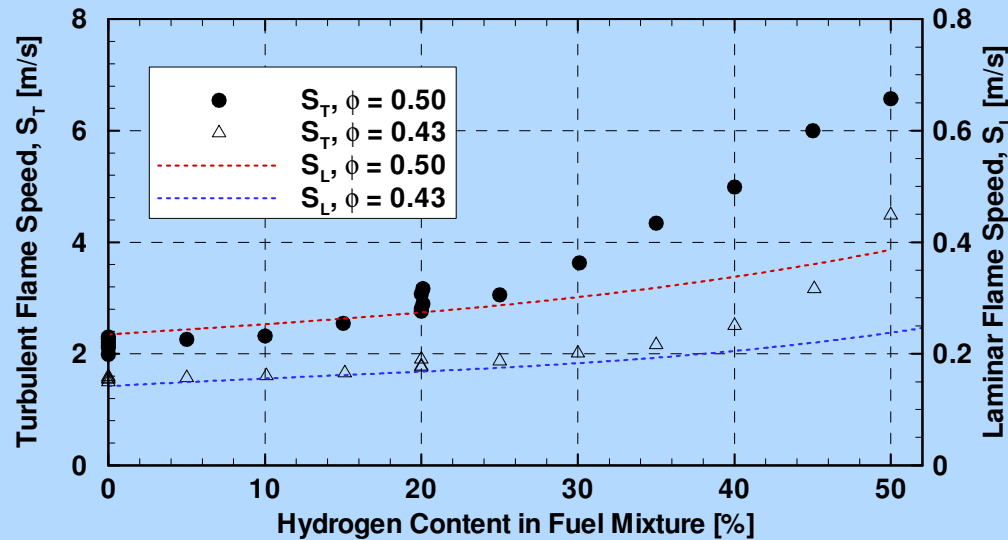
Turbulent Flame Speed

673 K, 5 bar, 40 m/s, grid g365,xg10



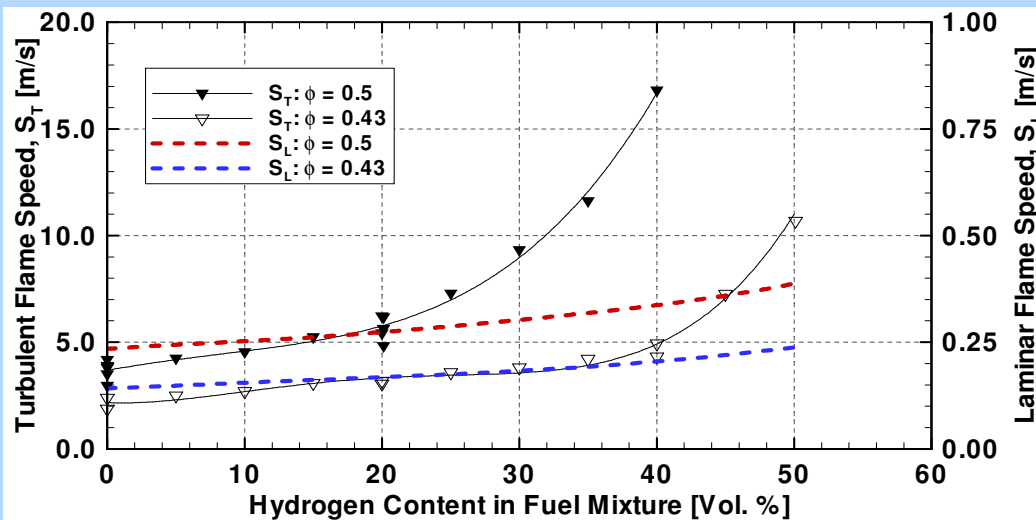
Up to approx. 25 % Vol. H_2 : chemical kinetics dominate (S_T/S_L const.)
 $\text{H}_2 > 25\%$ Vol.: additional effects (preferential diffusion, stretch)

Turbulent Flame Speed



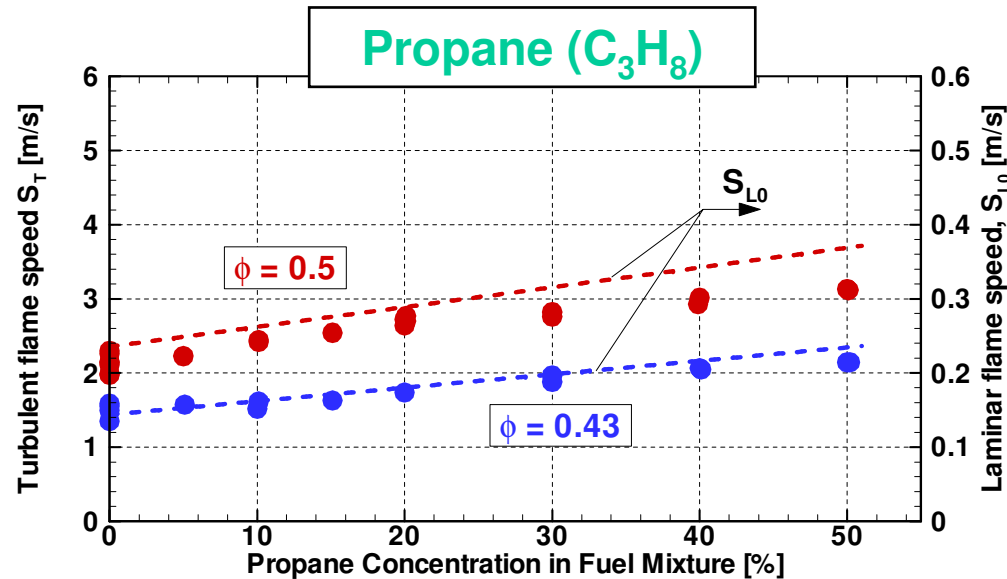
Low Turbulence Grid,
g350,xg30
Pre-Heating: $T = 673$ K
Bulk Velocity: $U_{\text{bulk}} = 40$ m/s
Pressure = 5 bars

**$S_T/S_L \neq \text{const.}$
 $= f(\text{H}_2 \text{ content})$
for $\text{H}_2 > 25\%$**



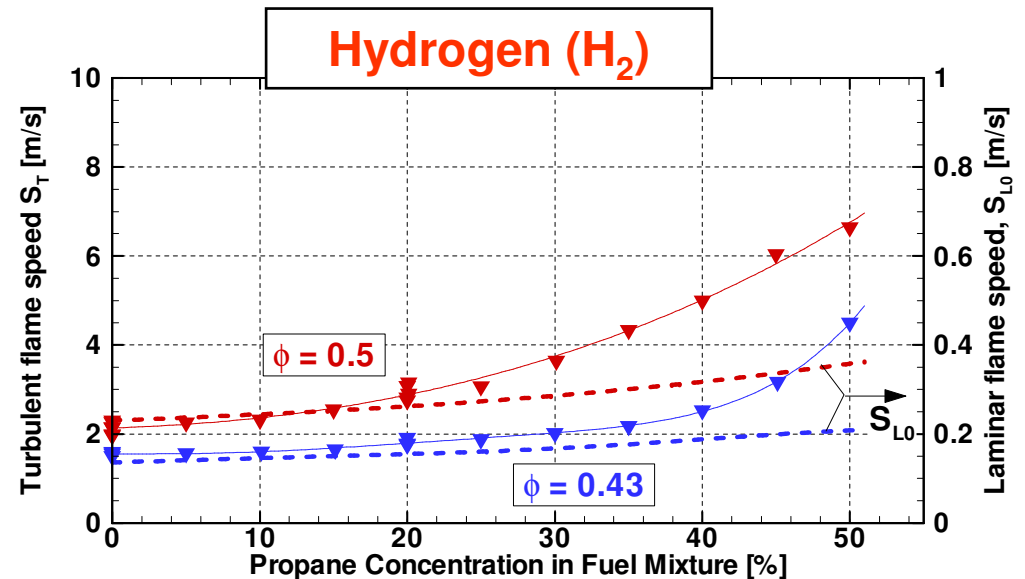
High Turbulence Grid,
g350,xg30
Pre-Heating: $T = 673$ K
Bulk Velocity: $U_{\text{bulk}} = 40$ m/s
Pressure = 5 bars

Turbulent Flame Speeds: Influence of Fuel Blend



- Effect of C_3H_8 addition:
 - Trend is consistent for both
 - Turbulent flame speed trend seems to be dominated by chemistry
 - Constant $S_T/S_L \approx 10$ in the measured range (100 % CH_4 – 50 % C_3H_8)

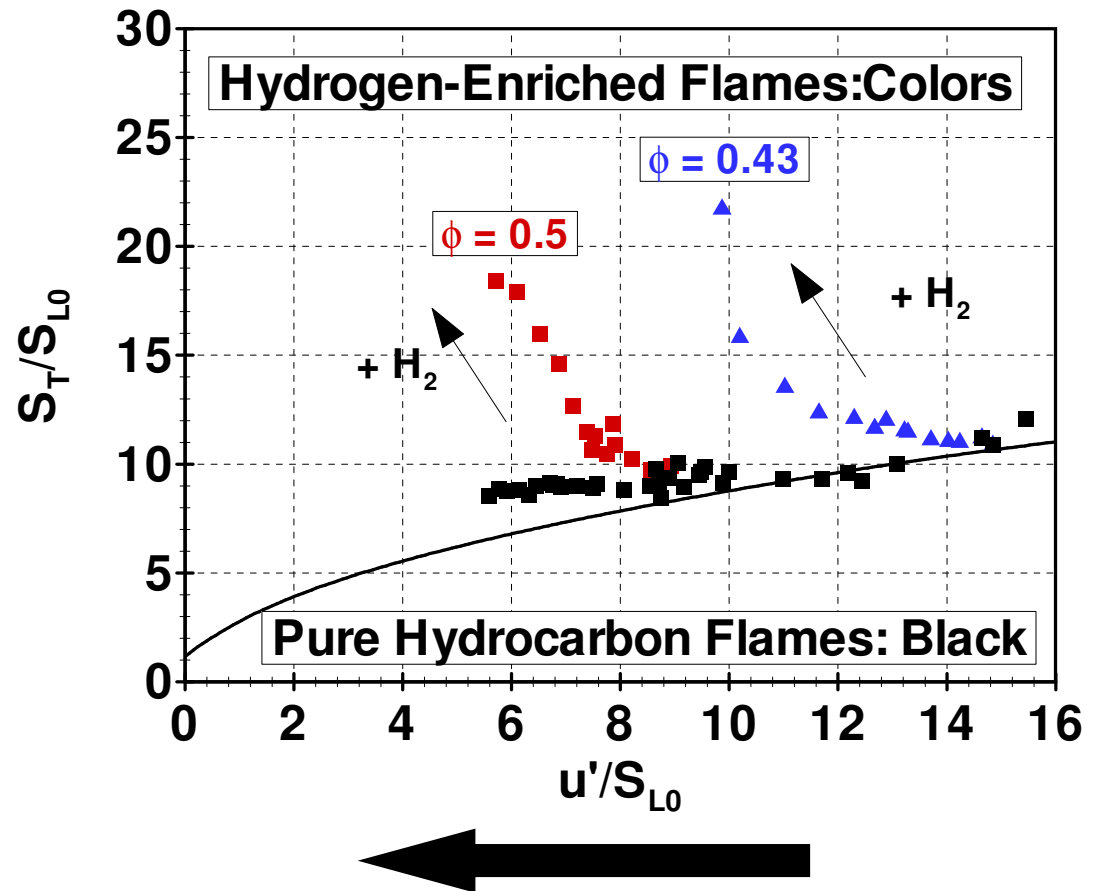
- Effect of H_2 addition:
 - S_T deviates greatly from S_L trend, especially at higher hydrogen content. i.e. more than just chemical effect
 - Effect is less pronounced for leaner mixtures.



Turbulent Flame Speeds

- Effect of turbulence intensity:

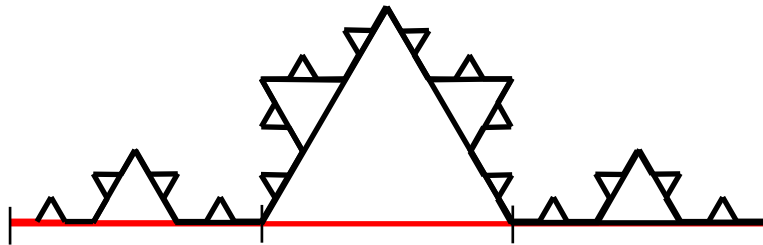
1. Generally, increasing turbulence level (u'/S_L) increases the enhancement of flame speed (S_T/S_L) but reaches asymptotic maximum (onset of quenching)
2. Hydrocarbon flames (methane and propane) fit well with this trend
3. Hydrogen enrichment exhibits a fundamentally different behavior – significant influence of stretch effects on laminar flame speed require correction terms



Unexpected Result: Decreasing
Turbulence Level increases $S_T/S_{L,0}$

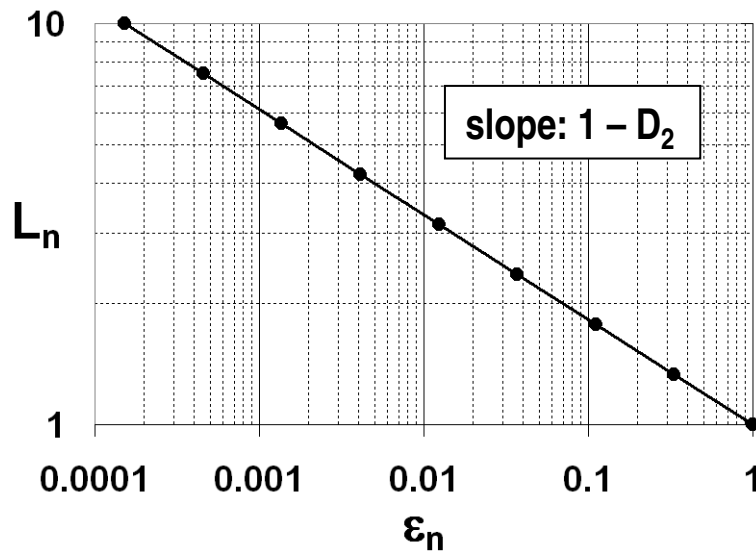
Fractals

self-similar geometry \rightarrow fractal



ϵ_n – length scale

L_n – length of a curve



relation between scale (ϵ_n) and length (L_n)

$$L_n \sim \epsilon_n^{(1-D_2)}$$

D_2 - fractal dimension of a curve

D_2 is a measure of corrugation

Smooth curve: $D_2 = 1$

Koch curve: $D_2 = 1.26$

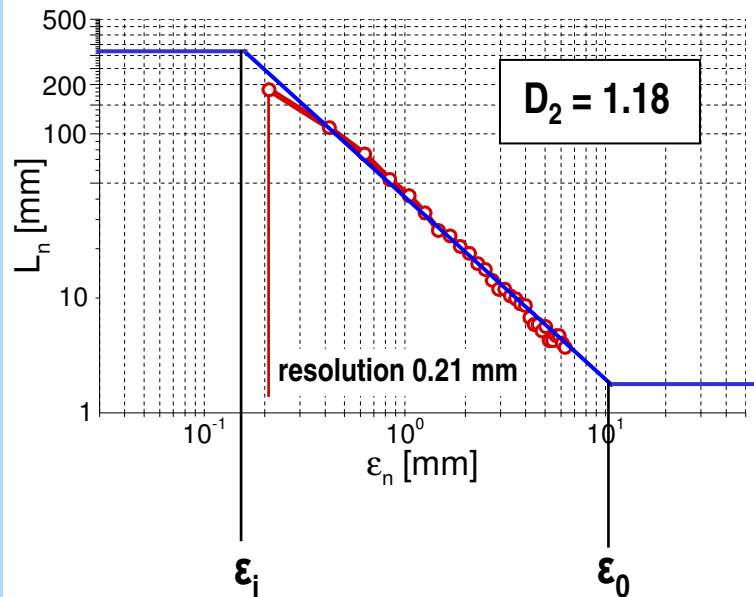
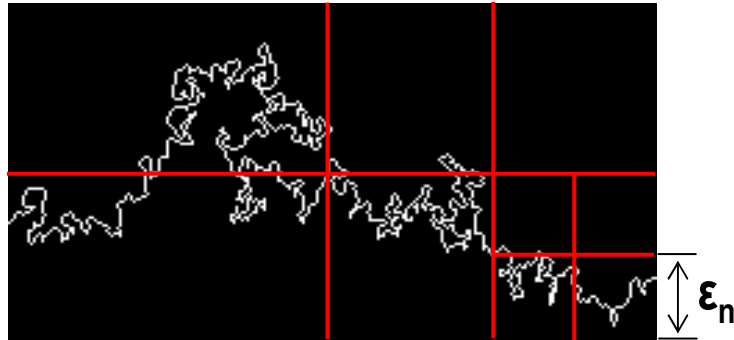
length L_n and surface A_n relation

$$A_n \sim \epsilon_n^{(1-D_3)}$$

$$D_3 = D_2 + 1$$

Fractal Nature of the Flame Front

fractal dimension of a flame front



inner and outer cut-off scales

outer cut-off scale (ϵ_0)

the largest scale \rightarrow Integral Length Scale (L_T)

inner cut-off scale (ϵ_i)

the smallest scale able to wrinkle the flame front

turbulent flame speed (s_T) – fractal approach

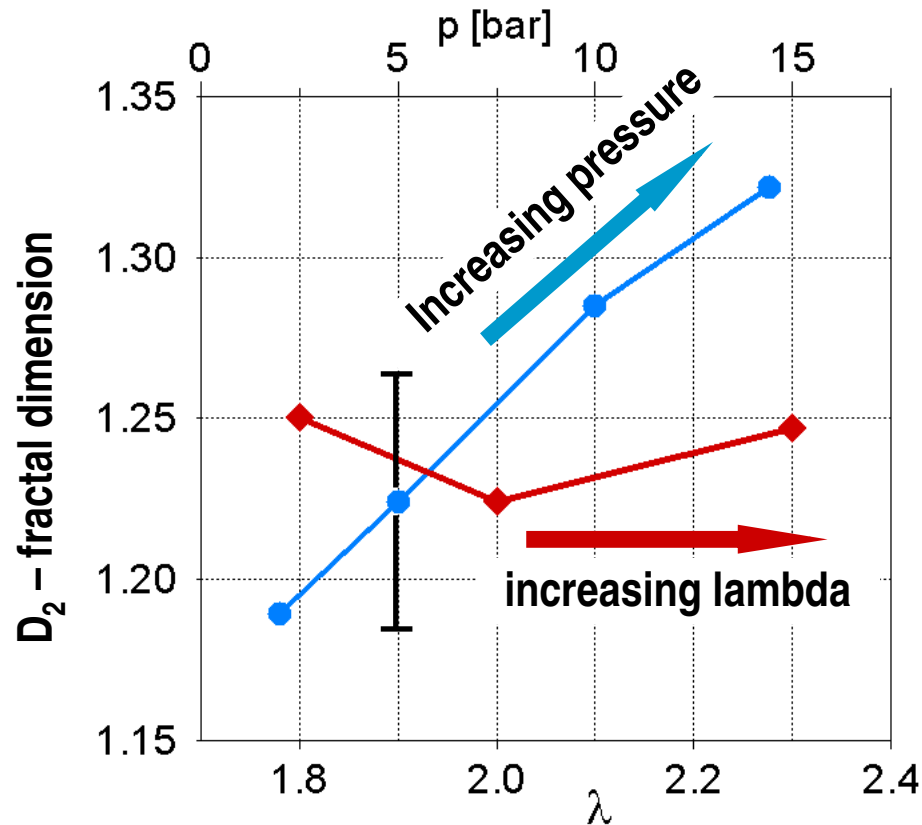
$$\frac{s_T}{s_L} = \frac{A_T}{A} = C \cdot \left(\frac{\epsilon_0}{\epsilon_i} \right)^{D_2 - 2} = \left(\frac{L_T}{\eta} \right)^{D_2 - 1}$$

Assumption:

inner cut-off scale is equal to Kolmogorov scale

$$L_T / \eta \sim Re_T^{3/4} \sim p^{3/4}$$

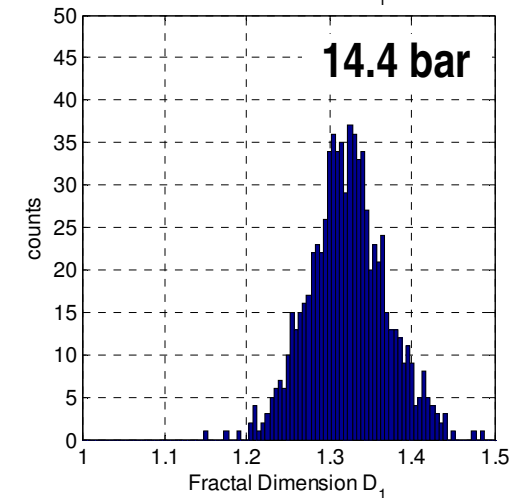
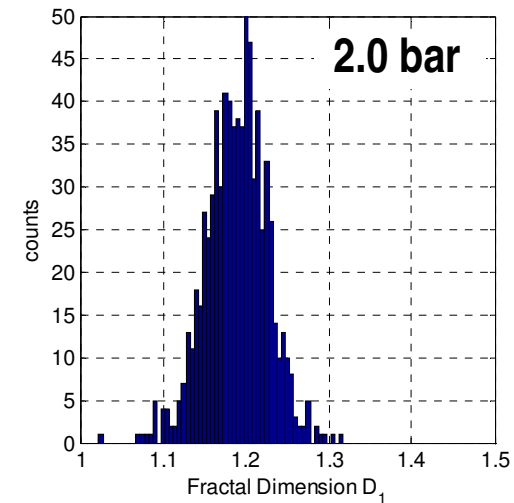
Fractals Dimension $D = f(p, \lambda)$



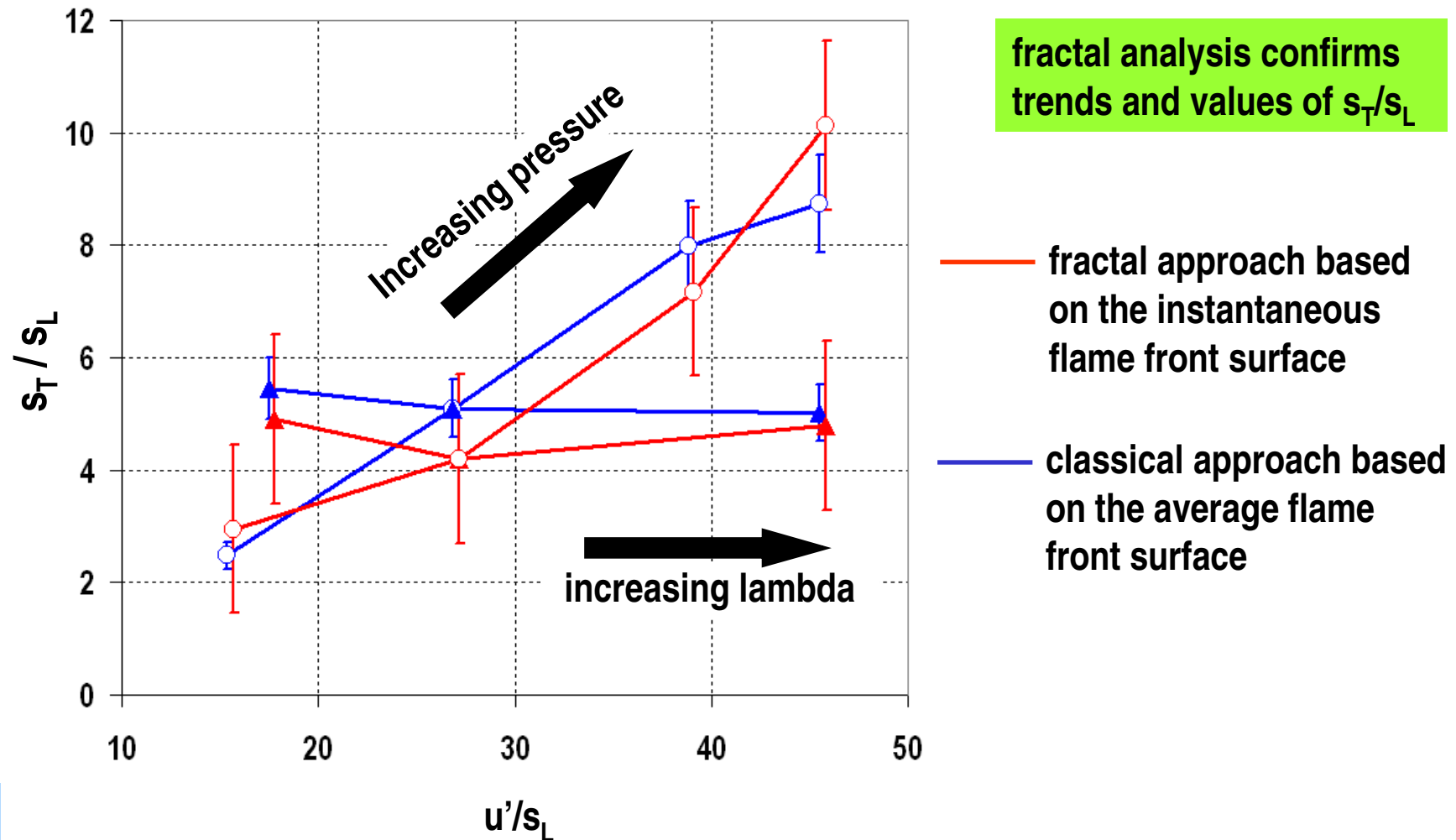
flame front corrugation (D_2)

- increases with pressure
- is independent of lambda

D_2 based on 800 single shots



s_T / s_L – Fractals Approach



1. Flame characterisation for a broad fuel spectrum at GT conditions (high-pressure test rig):

- **H₂- CH₄/air flames (ongoing work)**

determine fundamental reasons for specific characteristics
(turbulent flame speed at high H₂ content)

approach:

apply fractal analysis; derive stretch effects
(evaluate correction factor: Markstein length)

$$\frac{S_L}{S_{L,0}} = 1 - Ma \cdot Ka$$

- **Syngas (H₂/CO/inerts) combustion**

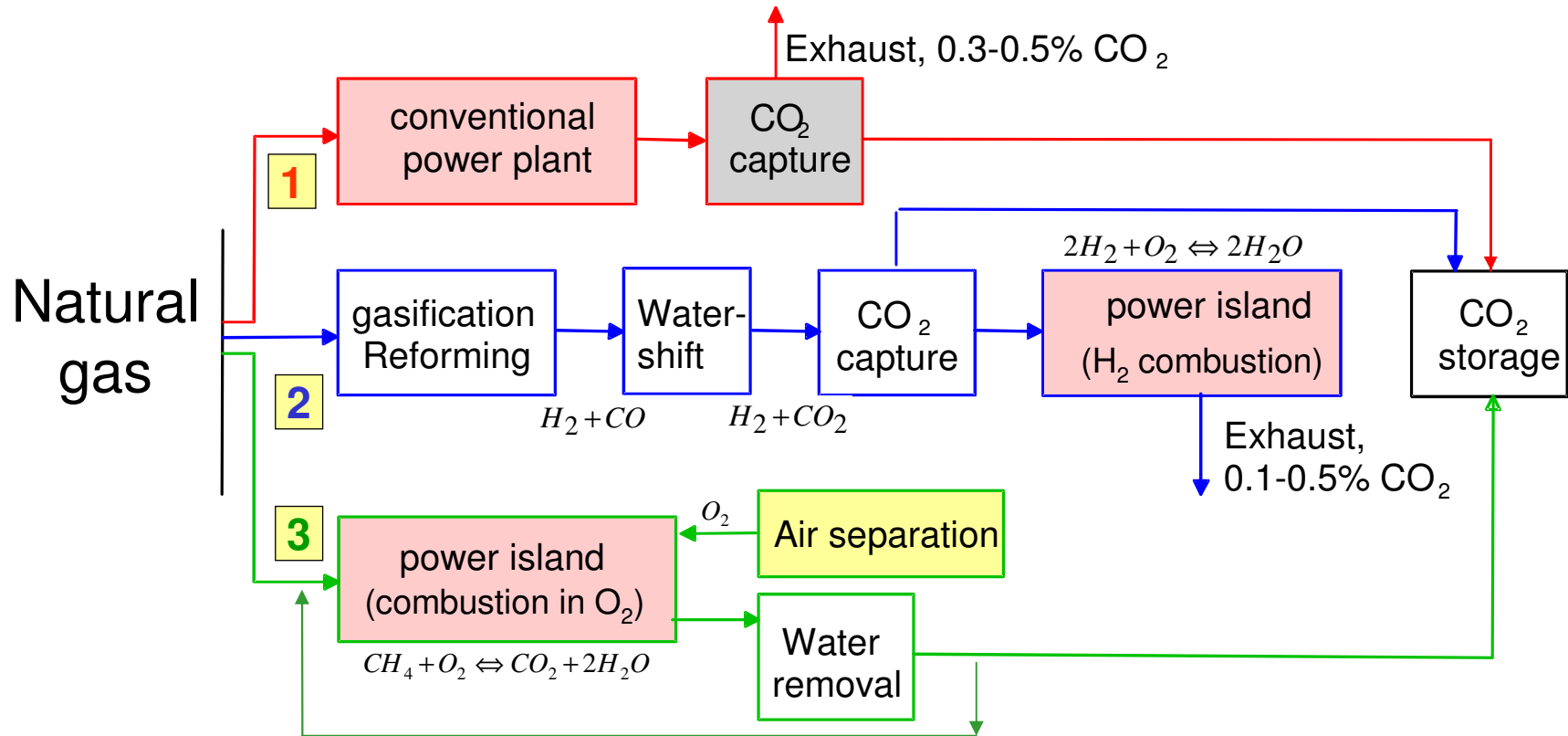
study flame characteristics (LBO/flashback, NO_x emission, flame speed)

motivation: demonstrate & understand safe, low-emission combustion
of fuel gases derived from gasification of solid fuels (biomass, coal, ...)

- **combustion of hydrogen rich fuel mixtures (> 80%Vol. H₂; rest: CO, inerts)**

study flame characteristics of fuel mixtures derived from pre-combustion
decarbonization process schemes

Gasturbine Technologies for CO₂-free power generation



1: Tail End Measure

2: Front End Measure

3: Integrated Measure: Oxy-fuel ($\lambda=1$; combustion in O₂)

Fuel Decarbonization

